Top Quality Water

Stanford Utilities Services is pleased to provide you with the 2013 Annual Water Quality Report. During 2013, the San Francisco Public Utilities Commission (SFPUC) and Stanford University monitored water quality for both source and treated water supplies, and in all cases the water quality was in compliance with the California Department of Public Health (CDPH) and the United States Environmental Protection Agency (USEPA) drinking water requirements. We continue our commitment to provide our customers with safe, high quality drinking water. It is the policy of Stanford Utilities Services to fully inform its consumers about the water quality standards and typical concentrations.

Utilities Services manages the storage, distribution, maintenance, and monitoring programs for Stanford’s drinking water supply. Stanford’s water supply is both chloraminated and fluoridated by SFPUC.

Stanford routinely collects water quality samples from various locations within the campus distribution system. The most frequently collected samples are analyzed for coliform bacteria, chloramine residual, and general physical parameters. Additional water quality samples are collected to monitor for more constituents in compliance with CDPH requirements. A California certified laboratory analyzes required samples. Stanford submits monthly reports to the CDPH that include all monitoring results.

SFPUC also collects daily water quality samples from various locations within their transmission system. The samples are analyzed for primary standards that apply to the protection of public health and secondary standards that refer to the aesthetic qualities of water, such as taste and odor.

Stanford Utilities Services also maintains flushing, cross-connection, and backflow prevention programs to ensure a consistent high quality drinking water supply.

In This Report

Stanford University's Drinking Water Sources 2
Contaminants in Drinking Water 3
Important Definitions 4
Water Quality Data 5
Additional Information about Water for Residents 6
Water Conservation for Residents 7
Contact Information 8
Stanford University’s Drinking Water Sources

Water supplied to Stanford by the SFPUC comes from three major sources: the Hetch Hetchy watershed, the Alameda watershed and the San Mateo watershed.

Hetch Hetchy Watershed

Hetch Hetchy Reservoir, is the largest reservoir in the SFPUC system, it is located in Yosemite National Park. In 2013, the Hetch Hetchy watershed provided the majority of the total water supply, with the remainder contributed by the two local watersheds. For the SFPUC system, the major source originates from spring snowmelt flowing down the Tuolumne River to the Hetch Hetchy Reservoir, where it is stored. This pristine Sierra water source meets all federal and state criteria for watershed protection. The SFPUC also maintains stringent disinfection treatment practices, extensive bacteriological quality monitoring, and high operational standards. As a result, the California Department of Public Health and USEPA have granted the Hetch Hetchy water source a filtration exemption. This exemption is contingent upon the Hetch Hetchy water quality continuing to meet all filtration avoidance criteria.

Alameda Watershed

The Alameda watershed, spans more than 35,000 acres in Alameda and Santa Clara Counties. Surface water from rainfall and runoff is collected in the Calaveras and San Antonio Reservoirs. Prior to distribution, water from the watershed is treated at the Sunol Valley Water Treatment Plant.

San Mateo Watershed

Surface water from rainfall and runoff captured in the 23,000 acre Peninsula watershed, located in San Mateo County, is stored in reservoirs, including Crystal Springs (Lower and Upper), San Andreas, and Pilarcitos. The water from these reservoirs is treated at the Harry Tracy Water Treatment Plant.

Watershed Protection

The SFPUC actively protects the water resources entrusted to its care. Hetch Hetchy watershed is surveyed annually to evaluate the sanitary conditions, water quality, potential contamination sources, and the results of watershed management activities conducted by SFPUC and its partner agencies (including National Park Service and US Forest Service). Once every five years the local watersheds and the approved standby water sources in Early Intake watershed, which includes Cherry Lake and Lake Eleanor, are surveyed. The latest five year survey was completed in 2011 for the period of 2006-2010. These surveys identified wildlife, stock, and human activities as potential contamination sources. The reports are available for review at the CDPH San Francisco District office, (510) 620-3474.
Contaminants in Drinking Water

In order to ensure that tap water is safe to drink, the USEPA and CDPH prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. CDPH regulations also establish limits for contaminants in bottled water that provide the same protection for public health.

Drinking water (including bottled water) may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the USEPA’s Safe Drinking Water Hotline (800) 426-4791.

The sources of drinking water (both tap and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals, which in some cases are radioactive. It can also pick up substances resulting from the presence of animals or from human activity. Such substances are called contaminants.

Contaminants that may be present in source water include:

**Inorganic Contaminants**, such as salts and metals, which can be naturally occurring or result from urban stormwater runoff, industrial or domestic wastewater discharge, oil and gas production, mining, or farming.

**Radioactive Contaminants**, can be naturally occurring or the result of oil and gas production and mining activities.

**Pesticides and Herbicides**, may originate from a variety of sources, such as agricultural, urban stormwater runoff, and residential uses.

**Organic Contaminants**, including synthetic and volatile organic compounds that are by-products of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application and septic systems.

**Microbiological Contaminants**, such as viruses and bacteria, may come from sewage treatment plants, septic systems, agricultural live stock operations, and wildlife. Many of these microorganisms are not visible to the naked eye and are not removed by standard treatment processes. Cryptosporidium

*Cryptosporidium* is a parasitic microbe found in most surface water. The SFPUC tests regularly for this water-borne pathogen, and found it at very low levels in source water and treated water in 2013. However, current test methods approved by the USEPA do not distinguish between dead organisms and those capable of causing disease. If ingested, these parasites may produce symptoms of nausea, stomach cramps, diarrhea, and associated headaches. *Cryptosporidium* must be ingested to cause disease, and it may be spread through means other than drinking water.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. USEPA / Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the USEPA’s Safe Drinking Water Hotline (800) 426-4791 or Website: www.epa.gov/safewater

Reducing Lead from Plumbing Fixtures

Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but cannot control the variety of materials used in plumbing components. If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead levels in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline (800) 426-4791, or at www.epa.gov/safewater/lead.
Important Definitions

The Water Quality Data table (Page 5) summarizes the 2013 detected drinking water contaminants and the information about their typical sources. An extensive water sample collection and testing protocol is used at the various water sources throughout the SFPUC transmission system and in the campus distribution system. Contaminants below detection limits are not shown, in accordance with CDPH guidance. The following are definitions of key terms noted on the adjacent water quality data table. These terms refer to the standards and goals for water quality.

**Maximum Contaminant Level (MCL):** The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs or MCLGs (see definitions below) as is economically and technologically feasible. Secondary MCLs (SMCLs) are set to protect the odor, taste, and appearance of drinking water.

**Maximum Contaminant Level Goal (MCLG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the USEPA.

**Public Health Goal (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Maximum Residual Disinfectant Level (MRDL):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**Maximum Residual Disinfectant Level Goal (MRDLG):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standard (PDWS):** MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**Regulatory Action Level (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Treatment Techniques (TT):** A required process intended to reduce the level of a contaminant in drinking water.

**Turbidity:** A water clarity indicator that is also used to indicate the effectiveness of the filtration system. High turbidity can hinder the effectiveness of disinfectants.

---

**Diverse Uses of Campus Domestic Water**

Swimming Pools  
Drinking Fountains  
Laboratories
### DETECTED CONTAMINANTS

#### Stanford University’s Annual Water Quality Data for 2013

<table>
<thead>
<tr>
<th>CONSTITUENTS WITH PRIMARY STANDARDS</th>
<th>Unit</th>
<th>MCL</th>
<th>PHG or (MCLG)</th>
<th>Range or Result</th>
<th>Average or (Maximum)</th>
<th>Typical Sources in Drinking Water</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>TURBIDITY (SFPUC samples)</strong></td>
<td>NTU</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Unfiltered Hetch Hetchy Water</td>
<td>NTU</td>
<td>5</td>
<td>N/A</td>
<td>0.2 - 0.3 (0)</td>
<td>(3.6) (0)</td>
<td>Soil runoff</td>
</tr>
<tr>
<td>Filtered Water - Sunol Valley WTP</td>
<td>NTU</td>
<td>1(4)</td>
<td>N/A</td>
<td>14 - 49</td>
<td>(50) (5)</td>
<td>By-product of drinking water chlorination</td>
</tr>
<tr>
<td>Minimum 95 % of samples &lt; 0.3 NTU (4)</td>
<td>-</td>
<td>-</td>
<td>N/A</td>
<td>100%</td>
<td>-</td>
<td>Soil runoff</td>
</tr>
<tr>
<td><strong>DISINFECTION BY-PRODUCTS</strong></td>
<td>ppm</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total Trihalomethanes (TTHMs) (Stanford samples)</td>
<td>ppm</td>
<td>80</td>
<td>N/A</td>
<td>37 - 62</td>
<td>(50) (5)</td>
<td>By-product of drinking water chlorination</td>
</tr>
<tr>
<td>Total Haloacetic Acids (HAAs) (Stanford samples)</td>
<td>ppm</td>
<td>60</td>
<td>N/A</td>
<td>14 - 49</td>
<td>(46) (5)</td>
<td>By-product of drinking water chlorination</td>
</tr>
<tr>
<td>Total Organic Carbon (TOC) (SFPUC samples)</td>
<td>ppm</td>
<td>TT</td>
<td>N/A</td>
<td>1 - 3.4</td>
<td>-</td>
<td>Various natural and man-made sources</td>
</tr>
<tr>
<td><strong>MICROBIOLOGICAL CONTAMINANTS</strong></td>
<td>%</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Total Coliform (Stanford samples)</td>
<td>%</td>
<td>≤5</td>
<td>(0)</td>
<td>-</td>
<td>(0)</td>
<td>Naturally present in the environment</td>
</tr>
<tr>
<td>Giardia Lamblia (SFPUC samples)</td>
<td>cyst/L</td>
<td>TT</td>
<td>(0)</td>
<td>&lt;0.01 - 0.04</td>
<td>&lt;0.01</td>
<td>Naturally present in the environment</td>
</tr>
<tr>
<td><strong>INORGANIC CONTAMINANTS</strong></td>
<td>ppm</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fluoride (source water) (SFPUC samples)</td>
<td>ppm</td>
<td>2.0</td>
<td>1.0</td>
<td>ND - 0.8 (7)</td>
<td>0.4 (7)</td>
<td>Erosion of natural deposits</td>
</tr>
<tr>
<td>Total Chlorine (Stanford samples)</td>
<td>ppm</td>
<td>MRDL=4</td>
<td>MRDLG=4</td>
<td>1.6-2.8</td>
<td>2.3 (8)</td>
<td>Water disinfectant added for treatment</td>
</tr>
<tr>
<td><strong>RADIOACTIVE CONTAMINANTS</strong></td>
<td>pCi/L</td>
<td>15</td>
<td>(0)</td>
<td>ND - 3.9</td>
<td>ND</td>
<td>Erosion of natural deposits</td>
</tr>
</tbody>
</table>

#### COUNSTITUENTS WITH SECONDARY STANDARDS

<table>
<thead>
<tr>
<th>CONSTITUENTS WITH SECONDARY STANDARDS</th>
<th>Unit</th>
<th>SMCL</th>
<th>PHG</th>
<th>Range</th>
<th>Average</th>
<th>Typical Sources in Drinking Water</th>
</tr>
</thead>
<tbody>
<tr>
<td>(SFPUC samples, except Color)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Aluminum</td>
<td>ppb</td>
<td>200</td>
<td>600</td>
<td>ND - 52</td>
<td>ND</td>
<td>Erosion of natural deposits</td>
</tr>
<tr>
<td>Chloride</td>
<td>ppm</td>
<td>500</td>
<td>N/A</td>
<td>&lt;3 - 18</td>
<td>10.2</td>
<td>Runoff / leaching from natural deposits</td>
</tr>
<tr>
<td>Color (Stanford samples)</td>
<td>unit</td>
<td>15</td>
<td>N/A</td>
<td>≤5 - 6</td>
<td>≤5</td>
<td>Naturally occurring organic materials</td>
</tr>
<tr>
<td>Specific Conductance</td>
<td>µS/cm</td>
<td>1600</td>
<td>N/A</td>
<td>29 - 258</td>
<td>169</td>
<td>Substances that form ions when in water</td>
</tr>
<tr>
<td>Sulfate</td>
<td>ppm</td>
<td>500</td>
<td>N/A</td>
<td>0.8 - 33</td>
<td>16.6</td>
<td>Runoff / leaching from natural deposits</td>
</tr>
<tr>
<td>Total Dissolved Solids</td>
<td>ppm</td>
<td>1000</td>
<td>N/A</td>
<td>&lt;20 - 109</td>
<td>71</td>
<td>Runoff / leaching from natural deposits</td>
</tr>
<tr>
<td>Turbidity</td>
<td>NTU</td>
<td>5</td>
<td>N/A</td>
<td>0.1 - 0.3</td>
<td>0.1</td>
<td>Soil runoff</td>
</tr>
</tbody>
</table>

#### LEAD AND COPPER

<table>
<thead>
<tr>
<th>LEAD AND COPPER</th>
<th>Unit</th>
<th>AL</th>
<th>PHG</th>
<th>Range</th>
<th>90th Percentile</th>
<th>Typical Sources in Drinking Water</th>
</tr>
</thead>
<tbody>
<tr>
<td>(Stanford Samples, 55 samples collected)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Copper</td>
<td>ppb</td>
<td>1300</td>
<td>300</td>
<td>&lt;50 - 100</td>
<td>73 (9)</td>
<td>Corrosion of household plumbing systems</td>
</tr>
<tr>
<td>Lead</td>
<td>ppb</td>
<td>15</td>
<td>0.2</td>
<td>&lt;5 - 10</td>
<td>&lt;5 (9)</td>
<td>Corrosion of household plumbing systems</td>
</tr>
</tbody>
</table>

#### OTHER WATER QUALITY PARAMETERS

<table>
<thead>
<tr>
<th>OTHER WATER QUALITY PARAMETERS (SFPUC Samples)</th>
<th>Unit</th>
<th>O RL</th>
<th>Range</th>
<th>Average</th>
</tr>
</thead>
<tbody>
<tr>
<td>Alkalinity (as CaCO₃)</td>
<td>ppm</td>
<td>N/A</td>
<td>7 - 71</td>
<td>46</td>
</tr>
<tr>
<td>Calcium (as Ca)</td>
<td>ppm</td>
<td>N/A</td>
<td>3 - 23</td>
<td>13</td>
</tr>
<tr>
<td>Chlorate (10)</td>
<td>ppb</td>
<td>800(NL)</td>
<td>39 - 690</td>
<td>303</td>
</tr>
<tr>
<td>Hardness (as CaCO₃)</td>
<td>ppm</td>
<td>N/A</td>
<td>7 - 89</td>
<td>53</td>
</tr>
<tr>
<td>Magnesium</td>
<td>ppm</td>
<td>N/A</td>
<td>0.2 - 8.3</td>
<td>5.3</td>
</tr>
<tr>
<td>pH</td>
<td>unit</td>
<td>N/A</td>
<td>6.5 - 9.4</td>
<td>8.4</td>
</tr>
<tr>
<td>Silica</td>
<td>ppm</td>
<td>N/A</td>
<td>4.8 - 6.2</td>
<td>5</td>
</tr>
<tr>
<td>Sodium</td>
<td>ppm</td>
<td>N/A</td>
<td>3 - 18</td>
<td>12</td>
</tr>
</tbody>
</table>

Key:
- AL = Action Level
- N/A = Not Applicable
- ND = Non-detect
- NL = Notification Level
- NTU = Nephelometric Turbidity Unit
- ORL = Other Regulatory Level
- ppb = parts per billion
- ppm = parts per million
- cyst/L = cyst per liter
- pCi/L = picocurie per liter
- µS/cm = microSiemens/centimeter
- TT = Treatment Technique

1. All results met State and Federal drinking water health standards.
2. Turbidity is measured every four hours. These are monthly average turbidity values.
3. The highest turbidity of the unfiltered water in 2013 was 3.6 NTU.
4. There is no turbidity MCL for filtered water. The limits are based on the TT requirements for filtration systems.
5. This is the highest locational, running, annual average.
6. Total organic carbon is a precursor for disinfection byproduct formation. The TT requirement applies to the filtered water from the SVWTP only.
7. The SFPUC adds fluoride to an optimum level of 0.9 ppm to help prevent dental caries in consumers. The CDPH specifies the fluoride levels in treated water must be maintained within a range of 0.8 ppm - 1.5 ppm. In 2013, the range and average of the fluoride levels were 0.7 ppm - 1.4 ppm and 0.9 ppm, respectively. The natural fluoride level in the Hetch Hetchy supply was ND.
8. This is the highest running annual average value.
9. The most recent lead and copper monitoring was in 2012. All results for samples collected at consumer taps were below the lead and copper Action Levels. Customer tap sampling is required again in 2015.
10. The detected chlorate in treated water is a degradation byproduct of sodium hypochlorite, used by the SFPUC for water disinfection.
Chloramine Degradation of Rubber Parts
Stanford University purchases 100% of its domestic water (disinfected with chloramines) from the SFPUC. Chloramines provide longer lasting water disinfection while complying with more stringent regulatory requirements for disinfection byproducts. Chloramines can cause faster breakdown of rubber parts used in household plumbing than the previously used disinfectant—chlorine. Typical household parts that can be affected by chloramines are: Stainless steel braided flex connectors. Rubber flapper valves in toilets.

The degradation can cause soft, black flakes in the water, usually on the first draw from the tap, but then the water quickly clears up. These flakes sometimes have a “greasy” appearance and will frequently stick and smear on surfaces.

To solve this problem, the degraded part should be replaced with new chloramine resistant materials (silicone-based rubber, synthetic polymers, copper, or nylon flex lines). These products should say “chloramine resistant”.

Emergency Preparedness
In the event of an earthquake, will you have enough drinking water?

Although Stanford strives to ensure a reliable supply of water for our customers, a natural disaster such as a major earthquake could interrupt water delivery. Residents are encouraged to store drinking water in case of an emergency.

Keep a 3-day water supply, just in case
- Each family member (including pets!) needs 1 gallon per day
- Store tap water in food-grade plastic containers; replace every 6 months
- Store bottled water in the original sealed containers; replace every 6 months

If your emergency supply runs out, you can treat your tap water as required or if notified
- Boil it for 3 minutes, or disinfect it by adding regular household bleach
- Add 8 drops of bleach per gallon of water
- Shake or stir then let it stand for 30 minutes

To learn more about emergency preparedness for yourself and your family, visit http://lbre.stanford.edu/sem/drinkingwater or http://www.72hours.org.
Stanford is a long-time partner with the Santa Clara Valley Water District (SCVWD) in offering residents valuable incentives for water conservation products, programs, and educational opportunities.

The following are some of the programs that are currently available. See our website for additional information: http://lbre.stanford.edu/sem/Water_Efficiency

Water Wise House Call
Contact the Santa Clara Valley Water District to schedule your FREE Water Wise House Call at: (800) 548-1882.

A SCVWD representative will meet with you to review your home water use and identify where you can improve efficiency. They will tell you about all of the conservation rebate programs available for your residence.

High Efficiency Toilet Rebate
You can receive a rebate by replacing toilets that use 1.6 gallons per flush or more. Up to three toilets per household are eligible.

Free Water Saving Devices
Receive FREE showerheads (2.0 gpm), faucet aerators for your kitchen and bathroom (2.2 and 1.5 gpm, respectively), shower timers and toilet leak detection tablets. Contact Jennifer Fitch at (650) 723-3494 or jcfitch@stanford.edu

Did you know that irrigation of an average landscape accounts for 60% of total single family residential household water use at Stanford?

Landscape Rebate Program
Replace your lawn with drought tolerant plantings and receive up to $2 per square foot of lawn replaced with either mulch or climate appropriate plants. (You must have a Water Wise House Call completed prior to being eligible for this rebate).

Weather-Based (Smart) Irrigation Controller Pilot Study & Rebate
Stanford’s Water Efficiency Program is looking for 50 single-family campus residents to participate in a pilot study to test weather-based (smart) irrigation controller technology. Visit our website to see if you qualify and for more details: http://lbre.stanford.edu/sem/Water_Efficiency

Faculty/Staff 2013 Water Use & 2014 Target

See graph below that shows faculty/staff water use for 2013 (purple) and 15% reduction target for 2014 (green).
Contact Information

USEPA Drinking Water Homepage:
http://www.epa.gov/safewater/ or Safe Drinking Water Hotline
(800) 426-4791

CDPH Drinking Water Program Homepage:
http://www.cdph.ca.gov/certlic/drinkingwater

SFPUC's Homepage:
http://sfwater.org

Stanford’s Utilities Water Homepage:
http://lbre.stanford.edu/sem/drinkingwater

If you have questions or need additional information about this report or Stanford's water quality, please contact:

Julia Nussbaum (650) 723-9747
E-mail: juliann@stanford.edu

Marty Laporte (650) 725-7864
E-mail: martyl@bonair.stanford.edu

Jennifer Fitch (650) 723-3494
E-mail: jcfitch@stanford.edu

Este reporte contiene información muy importante sobre el agua que toma. Llame a Stanford University (650) 725-8030 si necesita ayuda en español.